

## 1,3-Dipolar Character of Six-membered Aromatic Rings. Part 43.<sup>1</sup> Cycloadditions leading to Tricyclic Adducts

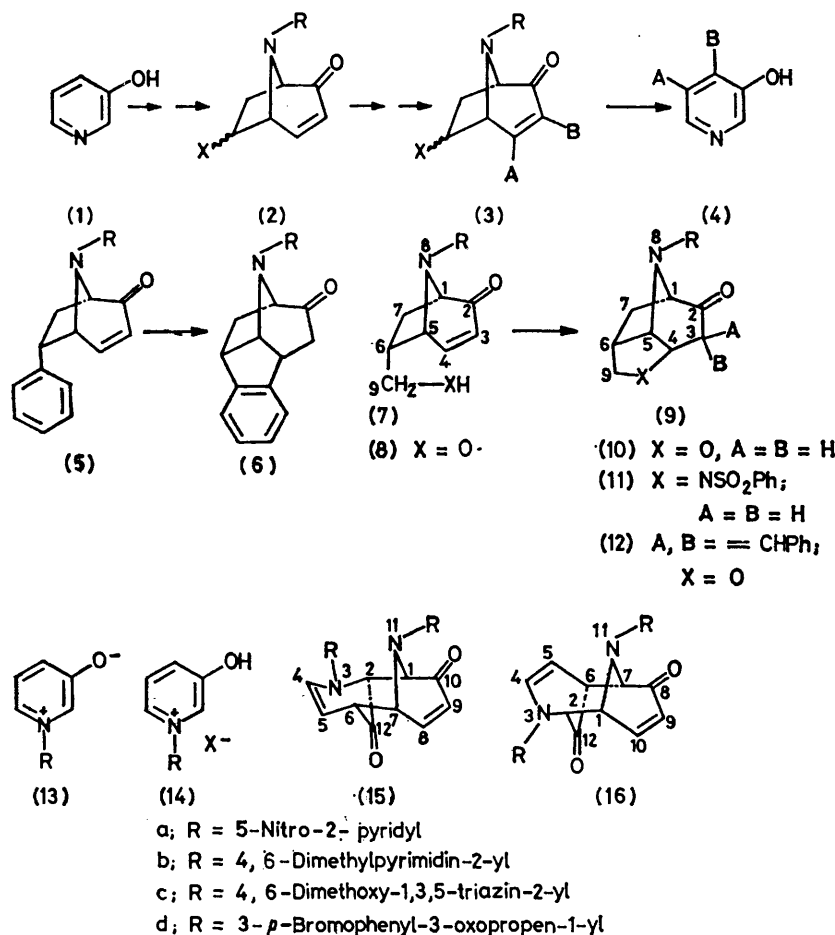
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Allyl alcohol reacts with 1-heteroaryl-3-oxidopyridiniums to give a tricyclic product (10) in which the OH group has added to the  $\alpha\beta$ -unsaturated ketone. Analogous products are obtained from *N*-allylbenzenesulphonamide, acrylic acid, and 2-vinylpyridine. Further transformations of the primary adducts are described.

A MAJOR aim of our cycloaddition work is the development of new synthetic methods in pyridine chemistry. Thus, the reversibility of cycloadditions suggests the conversion of cycloadducts (2) into derivatives (3) which should allow the substitution of 3-hydroxypyridine in the

derivatives (6) suggested that suitable *endo*-adducts of type (7) should undergo spontaneous cyclisation to the tetracycles (9). The present paper describes these and related systems.

The betaines used in this study (13a—d) were gener-



4- and/or 5-positions [(1)→(4)]. We have described the preparation of 4-bromo-3-hydroxypyridine by this method.<sup>2</sup> However, attempts to change the OH function of 3-hydroxypyridine by an analogous route are hindered by the preferential reaction<sup>2</sup> of nucleophiles with the C=C bond in (2). We therefore examined methods of protecting this double bond. The facile conversion<sup>3</sup> of styrene adducts (5) into the tetracyclic

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ated *in situ* either from their salts (14a—d) or from their dimers. Each betaine formed initially the *syn*-dimer (15a—d), but in the pyrimidinyl case, (15b) equilibrated to give mainly (16b).

### RESULTS AND DISCUSSION

*Allyl Alcohol.*—Whereas the pyridyl betaine (13a) reacted with allyl acetate to give the normal cycloadduct (17a), shown to have the *endo* structure by its n.m.r.

spectrum (Table 1), the reactions of betaines (13a), (13b), and (13c) with allyl alcohol each gave a product [(10a), (10b), and (10c) in 48%, 85%, and 10% yield, respectively] in which cyclisation of the intermediates (8a-c) had taken place. Structures (10) were proved by the n.m.r. spectra (see below) (Table 2), together with the i.r. spectra [saturated  $\nu(\text{C}=\text{O})$  at 1720–1730  $\text{cm}^{-1}$ ] and elemental analysis. The reaction with the pyrimidinyl dimer (16b) also yielded 14% of the *exo*-adduct (22b), in which cyclisation of the type (7)→(9) cannot occur. The poor yield obtained of (10c) is due to the

addition of allyl alcohol to the triazinyl dimer (15c) to give (24c). Similar products have been obtained previously from the pyridyl dimer (15a).<sup>4</sup> Structure (24c) was proved by the n.m.r. and i.r. spectrum: the  $\alpha\beta$ -unsaturated carbonyl band of the dimer (16c) was lost, the new saturated  $\nu(\text{C}=\text{O})$  evidently overlapping with the other saturated  $\nu(\text{C}=\text{O})$ . The <sup>1</sup>H n.m.r. spectrum showed the disappearance of the  $\alpha\beta$ -olefinic protons.

*Further Transformations of Allyl Alcohol Tricycles.*—The pyrimidinyl derivative (10b) reacted with phenylhydrazine to give the corresponding phenylhydrazone

TABLE 1

<sup>1</sup>H N.m.r. data (100 MHz) for adducts by addition at the 2,6-positions of the pyridine ring <sup>a</sup>

(a) Chemical shifts													
Proton	(17a) <sup>b</sup>	(18a) <sup>c</sup>	(19d) <sup>b</sup>	(20b) <sup>b</sup>	(21b) <sup>b</sup>	(22b) <sup>b</sup>	(23d) <sup>b</sup>	(37) <sup>b</sup>	(38) <sup>c</sup>	(40) <sup>b</sup>	(41) <sup>b</sup>	(42) <sup>b</sup>	(43) <sup>b</sup>
1	4.82 <sup>g</sup>	5.01 <sup>g</sup>	4.32 <sup>h</sup>	5.12 <sup>g</sup>	5.12 <sup>g</sup>	4.96 <sup>g</sup>	4.41 <sup>h</sup>	5.20 <sup>h</sup>	5.00 <sup>g</sup>	5.48 <sup>g</sup>	5.17 <sup>g</sup>	5.09 <sup>g</sup>	5.04 <sup>h</sup>
3	6.11 <sup>h</sup>	5.86 <sup>h</sup>	5.98 <sup>h</sup>	5.94 <sup>h</sup>	5.80 <sup>g</sup>	5.78 <sup>h</sup>	5.98 <sup>h</sup>	5.92 <sup>h</sup>	6.06 <sup>h</sup>	5.88 <sup>h</sup>	5.89 <sup>h</sup>	6.09 <sup>h</sup>	5.92 <sup>h</sup>
4	7.28 <sup>h</sup>	7.40 <sup>j</sup>	7.18 <sup>h</sup>	6.76 <sup>h</sup>	6.96 <sup>h</sup>	7.25 <sup>h</sup>	7.24 <sup>h</sup>	7.06 <sup>h</sup>	6.94 <sup>h</sup>	7.27 <sup>h</sup>	7.20 <sup>h</sup>	7.32 <sup>h</sup>	7.20 <sup>h</sup>
5	5.35 <sup>h</sup>	5.16 <sup>k</sup>	4.66 <sup>k</sup>	5.50 <sup>k</sup>	5.70 <sup>k</sup>	5.16 <sup>g</sup>	4.75 <sup>g</sup>	5.31 <sup>g</sup>	5.63 <sup>k</sup>	5.12 <sup>g</sup>	5.53 <sup>g</sup>	5.52 <sup>k</sup>	5.50 <sup>k</sup>
6	3.10 <sup>l</sup>	2.80 <sup>j</sup>	3.50 <sup>m</sup>	3.92 <sup>l</sup>	4.13 <sup>l</sup>	3.98 <sup>n</sup>	2.98 <sup>j</sup>	3.61 <sup>h</sup>	o	3.02 <sup>h</sup>	3.06 <sup>h</sup>	3.36 <sup>m</sup>	3.54 <sup>j</sup>
7- <i>exo</i>	2.78 <sup>n</sup>	2.56 <sup>j</sup>	2.70 <sup>n</sup>	2.92 <sup>n</sup>	2.86 <sup>l</sup>	2.20 <sup>j</sup>	2.84 <sup>j</sup>	2.76 <sup>l</sup>	o	2.74 <sup>n</sup>	2.79 <sup>n</sup>	2.90 <sup>n</sup>	2.70 <sup>l</sup>
7- <i>endo</i>	1.56 <sup>h</sup>	1.60 <sup>h</sup>	2.13 <sup>h</sup>	2.02 <sup>h</sup>	2.86 <sup>l</sup>	1.89 <sup>h</sup>	2.06 <sup>h</sup>	2.30 <sup>j</sup>	o	2.00 <sup>h</sup>	2.25 <sup>j</sup>	2.01 <sup>h</sup>	2.12 <sup>h</sup>
Aromatic		6.7—	7.5—	6.3—	6.3—	6.27 <sup>l</sup>	7.4—	6.3—	6.6—	6.27 <sup>l</sup>	6.38 <sup>l</sup>	6.38 <sup>l</sup>	6.38 <sup>l</sup>
CMe		8.9 <sup>l</sup>	7.7 <sup>l</sup>	8.5 <sup>l</sup>	8.5 <sup>l</sup>		7.5 <sup>j</sup>	8.5 <sup>l</sup>	9.1 <sup>l</sup>				
CH <sub>2</sub>	4.30 <sup>h</sup>	2.56 <sup>j</sup>	1.45 <sup>i</sup>	2.25 <sup>i</sup>	2.25 <sup>i</sup>	2.20 <sup>i</sup>	1.45 <sup>i</sup>	2.22 <sup>i</sup>	2.25 <sup>i</sup>	2.25 <sup>i</sup>	2.25 <sup>i</sup>	2.25 <sup>i</sup>	2.27 <sup>i</sup>
OMe	3.88 <sup>h</sup>	2.80 <sup>j</sup>				3.60 <sup>h</sup>							
N-Me	2.15 <sup>i</sup>					3.40 <sup>h</sup>							3.68 <sup>i</sup>
								4.49 <sup>i</sup>					
Proton	(44) <sup>b</sup>	(45) <sup>b</sup>	(46) <sup>b</sup>	(47) <sup>b</sup>	(48) <sup>b</sup>	(49) <sup>b</sup>	(50) <sup>d</sup>	(51) <sup>b</sup>	(52) <sup>b</sup>	(53) <sup>b</sup>	(54) <sup>b,e</sup>	(55) <sup>b,f</sup>	
1	5.14 <sup>h</sup>	4.92 <sup>g</sup>	4.96 <sup>g</sup>	5.00 <sup>g</sup>	5.12 <sup>h</sup>	5.06 <sup>h</sup>	4.80 <sup>h</sup>	5.26 <sup>l</sup>	5.58 <sup>g</sup>	5.10 <sup>j</sup>	5.62 <sup>i</sup>	5.62 <sup>i</sup>	
3	5.89 <sup>h</sup>	6.05 <sup>h</sup>	6.07 <sup>h</sup>	5.98 <sup>g</sup>	6.08 <sup>h</sup>	5.99 <sup>h</sup>	5.83 <sup>h</sup>	5.88 <sup>h</sup>	5.90 <sup>h</sup>	5.89 <sup>h</sup>	5.50 <sup>j</sup>	5.66 <sup>j</sup>	
4	7.34 <sup>h</sup>	7.18 <sup>h</sup>	7.14 <sup>h</sup>	7.19 <sup>h</sup>	7.16 <sup>h</sup>	7.08 <sup>h</sup>	7.02 <sup>h</sup>	7.18 <sup>h</sup>	7.39 <sup>h</sup>	7.26 <sup>h</sup>	7.55 <sup>j</sup>	7.20—7.60 <sup>j</sup>	
5	5.55 <sup>g</sup>	5.41 <sup>k</sup>	5.57 <sup>k</sup>	5.60 <sup>k</sup>	5.76 <sup>g</sup>	5.42 <sup>g</sup>	5.35 <sup>g</sup>	5.58 <sup>k</sup>	5.45 <sup>g</sup>	5.22 <sup>j</sup>	5.71 <sup>g</sup>	5.83 <sup>g</sup>	
6	2.96 <sup>j</sup>	4.49 <sup>n</sup>	5.43 <sup>n</sup>	4.84 <sup>j</sup>				3.32 <sup>g</sup>	4.00 <sup>i</sup>	2.60 <sup>l</sup>			
7- <i>exo</i>	2.96 <sup>j</sup>	2.84 <sup>m</sup>	2.98 <sup>m</sup>	2.70 <sup>l</sup>	3.49 <sup>h</sup>	3.10 <sup>h</sup>	2.56 <sup>h</sup>		3.49 <sup>g</sup>	2.50 <sup>l</sup>			
7- <i>endo</i>	2.02 <sup>h</sup>	1.65 <sup>h</sup>	1.67 <sup>h</sup>	1.85 <sup>j</sup>	2.38 <sup>g</sup>	1.76 <sup>g</sup>	2.02 <sup>j</sup>	4.00 <sup>i</sup>					
Aromatic	6.38 <sup>i</sup>	6.36 <sup>i</sup>	6.39 <sup>i</sup>	6.32 <sup>i</sup>	6.40 <sup>i</sup>	6.38 <sup>i</sup>	6.29 <sup>i</sup>	6.38 <sup>i</sup>	6.36 <sup>i</sup>	6.32 <sup>i</sup>			
CMe	2.27 <sup>i</sup>	1.20 <sup>p</sup>	2.25 <sup>i</sup>	2.25 <sup>i</sup>	2.25 <sup>i</sup>	2.25 <sup>i</sup>	1.74 <sup>i</sup>	1.22 <sup>k</sup>	1.22 <sup>k</sup>	2.24 <sup>i</sup>	1.26 <sup>k</sup>	1.32 <sup>k</sup>	
CH <sub>2</sub>		2.25 <sup>i</sup>				1.50 <sup>i</sup>	2.25 <sup>i</sup>		2.22 <sup>i</sup>	2.22 <sup>i</sup>			
OMe	3.68 <sup>i</sup>		2.04 <sup>i</sup>				1.94 <sup>i</sup>		4.10 <sup>p</sup>	4.10 <sup>p</sup>	1.20—2.00 <sup>j</sup>	4.24 <sup>p</sup>	4.29 <sup>p</sup>
(b) Coupling constants (Hz)													
	(17a)	(18a)	(19d)	(20b)	(21b)	(22b)	(23d)	(37)	(38)	(40)	(41)	(42)	(43)
<i>J</i> (1,3)	1	1	2	1	0.5	2	2	1		1	1	1	2
<i>J</i> (1,7- <i>exo</i> )	8	7	7	8	8	10	8	8	8	5	8	8	8
<i>J</i> (3,4)	10	12	10	9	10	11	10	10	10	10	10	10	10
<i>J</i> (4,5)	5	5	5	6	6	5	5	5	5	8	5	6	5
<i>J</i> (5,6)	5	5	5	6	6				5			6	5
<i>J</i> (6,7- <i>exo</i> )	8	o	5	9	10	o	4	4	o	9	3	8	9
<i>J</i> (6,7- <i>endo</i> )	6	5	6	7	8	5	10	9	o	4	8	6	6
<i>J</i> (6,9A)	5	o				11							
<i>J</i> (6,9B)	10	o				8							
<i>J</i> (7- <i>exo</i> ,7- <i>endo</i> )	14	13	14	14	16	10	14	14	o	15	14	14	14
<i>J</i> (9A,9B)	11	o				12							
	(44)	(45)	(46)	(47)	(48)	(49)	(51)	(51)	(52)	(53)	(54)	(55)	
<i>J</i> (1,3)	2	2	1		1	1	2	1	1	1	1	1	
<i>J</i> (1,7- <i>exo</i> )	8	8	9	7	8	8	9	5	5	6			
<i>J</i> (3,4)	10	10	10	10	10	10	10	10	10	10	10	10	
<i>J</i> (4,5)	5	5	5	5	6	5	5	6	8	5	5	5	
<i>J</i> (5,6)	5	5	5	5	5			6	0.5	5			
<i>J</i> (6,7- <i>exo</i> )	o	10	9	9						o			
<i>J</i> (6,7- <i>endo</i> )	9	5	4	6						o			
<i>J</i> (7- <i>exo</i> ,7- <i>endo</i> )	14	14	14	14	14	14	14						
<i>J</i> (9A,9B)													

<sup>a</sup> In p.p.m. relative to SiMe<sub>4</sub> as internal standard. <sup>b</sup> In CDCl<sub>3</sub>. <sup>c</sup> In (CD<sub>3</sub>)<sub>2</sub>SO. <sup>d</sup> In CCl<sub>4</sub>. <sup>e</sup> Isomer, m.p. 155 °C. <sup>f</sup> Isomer, m.p. 162 °C. <sup>g</sup> Doublet. <sup>h</sup> Double doublet. <sup>i</sup> Singlet. <sup>j</sup> Overlap with other signals. <sup>k</sup> Triplet. <sup>l</sup> Multiplet. <sup>m</sup> Double triplet. <sup>n</sup> Quartet of doublets. <sup>o</sup> Not measurable due to signal overlap. <sup>p</sup> Quartet.

TABLE 2  
<sup>1</sup>H N.m.r. data (100 MHz) for tricyclic adducts <sup>a,b</sup>

(a) Chemical shifts									
Proton	(10a) <sup>c</sup>	(10b) <sup>c</sup>	(10c) <sup>c,d</sup>	(11a) <sup>c</sup>	(11b) <sup>c</sup>	(12b) <sup>c,e</sup>	(25) <sup>f,g</sup>	(26) <sup>c,h</sup>	(27) <sup>c,i</sup>
1	5.38 <sup>l</sup>	5.01 <sup>l</sup>	5.05 <sup>l</sup>	4.68 <sup>l</sup>	4.92 <sup>l</sup>	5.25 <sup>m</sup>	5.70 <sup>l</sup>	4.30 <sup>l</sup>	4.96 <sup>n</sup>
3	2.98 <sup>o</sup>	2.30 <sup>m</sup>	2.43 <sup>p</sup>	2.80 <sup>m</sup>	2.66 <sup>o</sup>		2.00— 2.50 <sup>m</sup>	4.05 <sup>l</sup>	2.00 <sup>m</sup>
4	5.32 <sup>m</sup>	4.70 <sup>r</sup>	4.74 <sup>r</sup>	4.27 <sup>q</sup>	4.24 <sup>q</sup>	5.25 <sup>m</sup>	4.50 <sup>s</sup>	4.90 <sup>n</sup>	4.40 <sup>m</sup>
5	5.80 <sup>n</sup>	5.50 <sup>n</sup>	5.54 <sup>n</sup>	5.14 <sup>n</sup>	5.18 <sup>n</sup>	5.58 <sup>n</sup>	5.34 <sup>n</sup>	5.02 <sup>n</sup>	5.28 <sup>n</sup>
6	3.46 <sup>s</sup>	2.94 <sup>s</sup>	3.00 <sup>s</sup>	2.90 <sup>m</sup>	2.84 <sup>m</sup>	3.14 <sup>s</sup>	3.40— 3.90 <sup>m</sup>	3.10 <sup>m</sup>	2.40— 2.80 <sup>m</sup>
7- <i>exo</i>	2.98 <sup>m</sup>	2.50 <sup>m</sup>	2.50 <sup>m</sup>	2.12 <sup>q</sup>		2.60 <sup>s</sup>	2.88 <sup>s</sup>	2.70 <sup>m</sup>	2.40— 2.80 <sup>m</sup>
7- <i>endo</i>	2.10 <sup>l</sup>	1.62 <sup>q</sup>	1.70 <sup>q</sup>	1.68 <sup>l</sup>	1.50 <sup>l</sup>	1.84 <sup>l</sup>	1.50 <sup>q</sup>	1.58 <sup>m</sup>	1.60 <sup>s</sup>
9- <i>exo</i>	4.20 <sup>s</sup>	3.92 <sup>q</sup>	3.70— 4.20 <sup>m</sup>	3.40 <sup>q</sup>	3.30 <sup>q</sup>	4.24 <sup>q</sup>	3.40— 3.90 <sup>m</sup>	3.50 <sup>q</sup>	3.67 <sup>q</sup>
9- <i>endo</i>	4.20 <sup>s</sup>	3.70 <sup>q</sup>	3.70— 4.70 <sup>m</sup>	3.16 <sup>q</sup>	3.12 <sup>q</sup>	3.77 <sup>q</sup>	3.40— 3.90 <sup>m</sup>	2.70 <sup>m</sup>	3.67 <sup>q</sup>
Proton	(28) <sup>c</sup>	(29) <sup>c</sup>	(30) <sup>q</sup>	(31) <sup>q</sup>	(32) <sup>c,j</sup>	(33) <sup>c,k</sup>	(35a) <sup>c</sup>	(35b) <sup>c</sup>	(36) <sup>c</sup>
1	5.00 <sup>l</sup>	5.65 <sup>l</sup>	5.49 <sup>l</sup>	5.64 <sup>l</sup>	6.00 <sup>l</sup>	5.43 <sup>l</sup>	5.40 <sup>l</sup>	5.08 <sup>l</sup>	5.00 <sup>l</sup>
3	2.20 <sup>m</sup>						3.02 <sup>o</sup>	2.40 <sup>q</sup>	2.75 <sup>l</sup>
4	4.74 <sup>r</sup>	5.04 <sup>l</sup>	4.84 <sup>l</sup>	4.89 <sup>l</sup>	6.53 <sup>l</sup>	6.47 <sup>l</sup>	5.68 <sup>s</sup>	5.22 <sup>r</sup>	6.00 <sup>m</sup>
5	5.50 <sup>n</sup>	5.84 <sup>n</sup>	5.65 <sup>n</sup>	5.84 <sup>n</sup>	5.62 <sup>n</sup>	5.56 <sup>n</sup>	6.10 <sup>n</sup>	5.78 <sup>n</sup>	6.00 <sup>m</sup>
6	2.94 <sup>s</sup>	3.00 <sup>s</sup>	3.00 <sup>m</sup>	2.90 <sup>m</sup>	3.30 <sup>s</sup>	3.12 <sup>s</sup>	3.74 <sup>r</sup>	3.14 <sup>r</sup>	4.50 <sup>s</sup>
7- <i>exo</i>	2.20 <sup>m</sup>	2.50 <sup>m</sup>	2.40 <sup>m</sup>	2.40 <sup>m</sup>	2.70 <sup>r</sup>	2.64 <sup>s</sup>	2.95 <sup>m</sup>	2.50 <sup>m</sup>	2.29 <sup>m</sup>
7- <i>endo</i>	1.68 <sup>q</sup>	1.66 <sup>q</sup>	1.50 <sup>q</sup>	1.49 <sup>q</sup>	1.95 <sup>l</sup>	1.82 <sup>l</sup>	2.34 <sup>q</sup>	1.98 <sup>q</sup>	1.86 <sup>q</sup>
9- <i>exo</i>	3.96 <sup>q</sup>	4.10 <sup>q</sup>	3.90 <sup>q</sup>	3.91 <sup>q</sup>	4.38 <sup>n</sup>	4.24 <sup>n</sup>			
9- <i>endo</i>	3.76 <sup>q</sup>	3.69 <sup>q</sup>	3.52 <sup>q</sup>	3.46 <sup>q</sup>	3.50 <sup>q</sup>	3.30 <sup>q</sup>			
(b) Coupling constants (Hz)									
	(10a)	(10b)	(10c)	(11a)	(11b)	(12b)	(25)	(26)	(27)
<i>J</i> (1,7- <i>exo</i> )	9	7	8	8	8	6	8	5	8
<i>J</i> (3- <i>exo</i> ,4)	<i>t</i>	2	<i>t</i>	6	5				
<i>J</i> (3- <i>endo</i> ,4)	<i>t</i>	4	<i>t</i>	1	1				
<i>J</i> (4,5)	7	7	6	7	6	7	7	8	7
<i>J</i> (5,6)	7	7	6	7	6	7	7	8	7
<i>J</i> (6,7- <i>exo</i> )	<i>t</i>	6	<i>t</i>	6	<i>t</i>	<i>t</i>	<i>t</i>	<i>t</i>	<i>t</i>
<i>J</i> (6,7- <i>endo</i> )	1	2	2		1	1	2	1	2
<i>J</i> (6,9- <i>exo</i> )	<i>t</i>	5	<i>t</i>	4	2	7	4	4	3
<i>J</i> (6,9- <i>endo</i> )	<i>t</i>	1	<i>t</i>	4	5	4	1	<i>t</i>	<i>t</i>
<i>J</i> (7- <i>exo</i> ,7- <i>endo</i> )	16	13	14	13	14	14	<i>t</i>	13	11
<i>J</i> (9- <i>exo</i> ,9- <i>endo</i> )	<i>t</i>	10	<i>t</i>	10	12	10	12	10	12
	(28)	(29)	(30)	(31)	(32)	(33)	(35a)	(35b)	(36)
<i>J</i> (1,7- <i>exo</i> )	6	6	7	7	6	7	8	8	8
<i>J</i> (3- <i>exo</i> ,4)	5						<i>t</i>	7	<i>t</i>
<i>J</i> (3- <i>endo</i> ,4)							<i>t</i>	2	<i>t</i>
<i>J</i> (4,5)	7	7	8	8	6	6	8	7	<i>t</i>
<i>J</i> (5,6)	7	7	8	8	6	6	8	7	<i>t</i>
<i>J</i> (6,7- <i>exo</i> )	<i>t</i>	<i>t</i>	<i>t</i>	<i>t</i>	8	<i>t</i>	10	10	<i>t</i>
<i>J</i> (6,7- <i>endo</i> )	2	3	3	2	1	1	2	2	3
<i>J</i> (6,9- <i>exo</i> )	5	6	6	6	9	8			
<i>J</i> (6,9- <i>endo</i> )	2	2	3	3	4	4			
<i>J</i> (7- <i>exo</i> ,7- <i>endo</i> )	13	14	14	14	14	14	16	14	14
<i>J</i> (9- <i>exo</i> ,9- <i>endo</i> )	10	10	10	10	9	10			

<sup>a</sup> In p.p.m. relative to SiMe<sub>4</sub> as internal standard. <sup>b</sup> Numbering is non-systematic, for comparison only. <sup>c</sup> In CDCl<sub>3</sub>. <sup>d</sup> OMe, 4.00 (footnote *p*). <sup>e</sup> C-CH<sub>3</sub>, 2.27 (footnote *p*). <sup>f</sup> NH, 9.11 (footnote *p*). <sup>g</sup> In (CD<sub>2</sub>)<sub>2</sub>SO. <sup>h</sup> 2''-H, 6''-H, 2.70 (footnote *m*); 3''-H, 5''-H, 3.10 (footnote *m*); CMe, 2.22 (footnote *p*). <sup>i</sup> 2-H, 4.40 (footnote *m*). <sup>j</sup> CMe, 2.20 (footnote *p*). <sup>k</sup> CMe, 2.15 (footnote *p*). <sup>l</sup> Doublet. <sup>m</sup> Overlapped with other signals. <sup>n</sup> Triplet. <sup>o</sup> Pseudo-singlet. <sup>p</sup> Singlet. <sup>q</sup> Doublet. <sup>r</sup> Quartet of doublets. <sup>s</sup> Multiplet. <sup>t</sup> Not measurable owing to signal overlap.

(25), but several attempts to achieve a Fischer indole synthesis <sup>5</sup> failed. Similarly, attempts at the Pfitzinger <sup>6</sup> and Friedlaender quinoline <sup>7</sup> synthesis failed. Cycloadduct (10b) condensed with benzaldehyde to the styryl derivative (12b) and with morpholine to form the enamine (26), which did not condense smoothly with methyl vinyl ketone. Borohydride reduction of (10b) gave the corresponding alcohol (27).

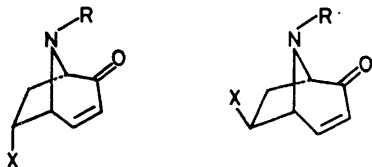
Under acidic conditions, bromination of (10b) occurred exclusively in the pyrimidine ring to give the *N*-(5-bromo-4,6-dimethylpyrimidin-2-yl) derivative (28). In pyridine solution, (10b) yielded the tribrominated compound (29) whereas the 2-pyridyl tricyclic adduct (10a)

gave a mixture of the dibromo- (30) and tribromo- (31) compounds. The value of  $\nu(\text{C}=\text{O})$  in (29)—(31) was raised by the  $\alpha$ -halogenation to 1 750—1 740 cm<sup>-1</sup>.

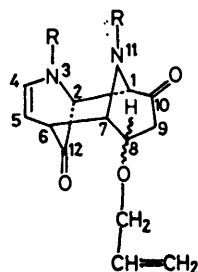
Benzofuroxan reacted with (10b) to form the quinoxaline 1,4-dioxide (32) in an example of the 'Bierut reaction'.<sup>8</sup> The dioxide (32) was stereospecifically reduced by KBH<sub>4</sub> to the monoxide (33). Structures (32) and (33) were established by spectroscopic evidence: in the i.r. spectrum no  $\nu(\text{C}=\text{O})$  or  $\nu(\text{OH}/\text{NH})$  was present; the aromatic ring-stretch showed at 1 580 cm<sup>-1</sup> and the *N*-oxide  $\nu(\text{N}-\text{O})$  at 1 360—1 370 cm<sup>-1</sup> (*cf.* ref. 9). This last band was considerably weaker in (33) than in (32).

*Allylamine and Derivatives.*—Allylamine itself, under

various conditions did not react with betaine (13b) but only with the dimer (16b) to give the dimer addition product (34), whereas with (13a) no homogeneous product

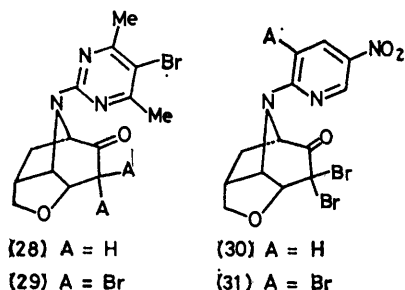
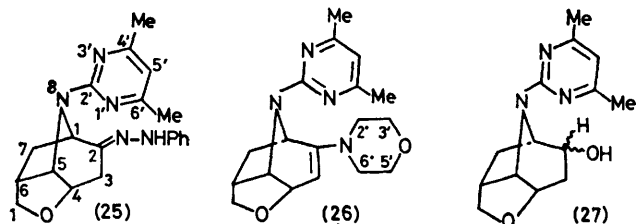


- (17) X = CH<sub>2</sub>OCOMe  
 (18) X = CH<sub>2</sub>NHSO<sub>2</sub>Ph  
 (19) X = CO<sub>2</sub>But  
 (20) X = 4-Pyridyl  
 (21) X = 2-Pyridyl



- (24) a; R = 5-Nitro-2-pyridyl  
 b; R = 4,6-Dimethylpyrimidin-2-yl  
 c; R = 4,6-Dimethoxy-1,3,5-triazin-2-yl  
 d; R = 3-*p*-Bromophenyl-3-oxopropen-1-yl

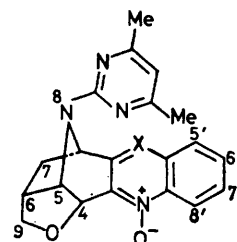
could be isolated. However, the cyclised products (11a and b) were obtained in 22 and 68% yield by the reaction of *N*-allylbenzenesulphonamide with the betaines (13a and b), respectively: again structures are supported by



n.m.r. spectra (see below) and i.r. spectra [ $\nu(\text{C}=\text{O})$  in Nujol at 1730 and 1720 cm<sup>-1</sup>]. The former reaction also afforded 20% of the uncyclised 6-*endo*-cycloadduct (18a). Surprisingly no cycloadducts were obtained from

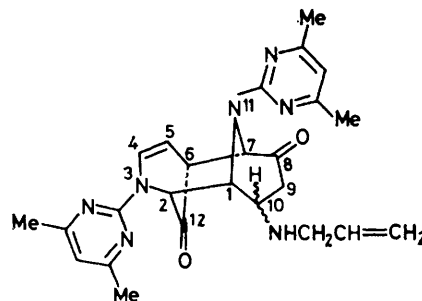
attempted reaction of *N*-alkylbenzenesulphonamide with (16c and d), nor from *N*-allylacetamide with (16a and b).

*Acrylic Acid Derivatives.*—Triethylammonium acrylate reacted with the pyridyl (13a) and pyrimidinyl (13b) betaines [but not with (16c or d)] forming the tricyclic



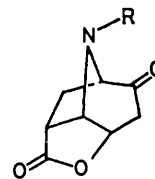
- (32) X = N<sup>+</sup>O<sup>-</sup>  
 (33) X = N

products (35a and b) in 44 and 52% yield, respectively. The i.r. spectra of both products showed the lactone  $\nu(\text{C}=\text{O})$  at 1770—1780 and the ketone  $\nu(\text{C}=\text{O})$  at 1720



(34)

cm<sup>-1</sup>. The n.m.r. spectra are discussed below. Betaines (13c and d) failed to react with acrylic acid.

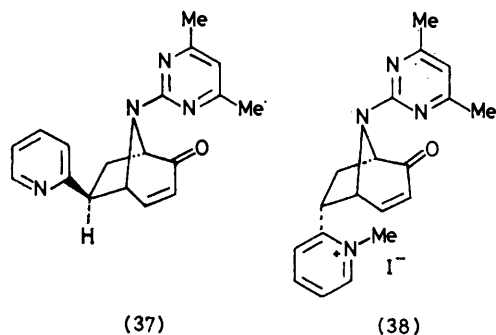
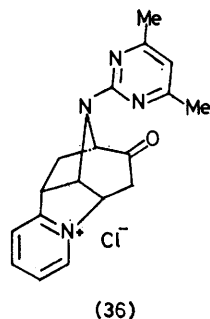


- (35) a; R = 5-Nitro-2-pyridyl  
 b; R = 4,6-Dimethylpyrimidin-2-yl

*t*-Butyl acrylate reacted with betaine (13d) to give a mixture of *endo*- (19d) and *exo*- (23d) adducts. It is interesting that the acrylate esters gave both stereoisomers, whereas from acrylic acid only products of the *endo*-addition were isolated.

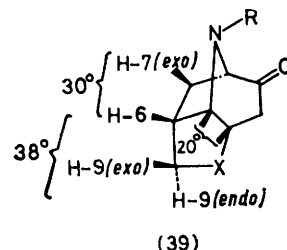
*Vinylpyridines.*—Betaine (13b) reacted normally with 4-vinylpyridine to give the *endo*-adduct (20b); the corresponding adduct (21b) from 2-vinylpyridine formed a cation which cyclised spontaneously to give (36). A low yield of the 6-*exo*-adduct (37) was also isolated. Reaction of (36) with aqueous base regenerated (21b). Quaternisation of adduct (21b) with methyl iodide gave a quaternary salt (38) ( $\delta$  4.49).

Hydrogen-1 N.M.R. Spectra of Tricyclic Adducts (Table 2).—The usual olefinic protons 3-H and 4-H were absent from the  $^1\text{H}$  n.m.r. spectra of these compounds.



The  $^1\text{H}$  n.m.r. spectra and decoupling experiments of the tricyclic compound (10b) are representative of this new class. Double irradiation at 5-H collapsed the 4-H multiplet at  $\delta$  4.70 into a broad singlet, and simplified

the 6-H multiplet at  $\delta$  2.94. Protons 6-H, 9-*exo*-H, and 9-*endo*-H constitute an ABX spin system in which the coupling of 5.0 Hz between 6-H and 9-*exo*-H correlates with a dihedral angle ( $\phi$ ) of  $38^\circ$  between 6-*exo*-H and 7-H. The coupling of 7.0 Hz between 4-H and 5-H corresponds to a dihedral angle  $^{10}$  of  $20^\circ$  [see (39)].



Other Cycloadditions at the 2,6-Positions of the Pyrimidinyl Betaine (13b).—Considerable further positive work is summarised in Table 3,<sup>9</sup> whereas attempted cycloadditions failed with the following dipolarophiles: phenyl isocyanate, phenyl isothiocyanate, benzalaniline, dicyclohexylcarbodi-imide, dihydropyran, allyl disulphide, and acrylamide.

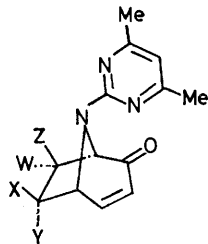
The electron-rich dipolarophiles, ethyl vinyl ether, vinyl acetate, and *N*-vinylpyrrolidin-2-one yielded exclusively the 6-*endo*-isomers (45), (46), and (47), respectively. In the case of isopropenyl acetate, it was not possible to determine the stereochemistry of the acetate group of the single isomer (50). As expected, the electron-deficient dipolarophiles acrylonitrile and methyl acrylate produced the 6-*endo*-6-*exo*-mixtures of isomers (42-41) and (43-44) respectively. Both  $\alpha$ -

TABLE 3  
Preparation of cycloadducts

Cyclo-adduct	Dipolarophile	Solvent	Vol (ml)	Reaction time/d	Yield (%)	M.p. ( $^\circ\text{C}$ )	Crystallisation solvent <sup>a</sup>	Analysis						
								Found			Formula	Required		
								C	H	N		C	H	N
(40)	Methyl vinyl ketone	MeCN	20	3	17	130-131	Et <sub>2</sub> O-EtOH	66.3	6.4	15.4	C <sub>15</sub> H <sub>17</sub> N <sub>3</sub> O <sub>2</sub>	66.4	6.3	15.5
(41)	Acrylonitrile	MeCN	15	4	44	206-207	EtOH	66.1	5.8	22.2	C <sub>14</sub> H <sub>14</sub> N <sub>4</sub> O	66.1	5.5	22.0
(42)	Acrylonitrile	MeCN	15	4	8	216-217	EtOH	66.1	5.7	22.3	C <sub>14</sub> H <sub>14</sub> N <sub>4</sub> O	66.1	5.5	22.0
(43)	Methyl acrylate	MeCN	10	2	40	110-111	Et <sub>2</sub> O-C <sub>6</sub> H <sub>12</sub>	63.0	6.2	14.4	C <sub>15</sub> H <sub>17</sub> N <sub>3</sub> O <sub>3</sub>	62.7	5.9	14.6
(44)	Methyl acrylate	MeCN	10	2	29	87-88	Et <sub>2</sub> O-C <sub>6</sub> H <sub>12</sub>	62.9	6.3	14.5	C <sub>15</sub> H <sub>17</sub> N <sub>3</sub> O <sub>3</sub>	62.7	5.9	14.6
(45)	Ethyl vinyl ether <sup>b</sup>	1,2-dichloroethane	20	4	41	126-127	EtOH	65.9	7.0	15.5	C <sub>15</sub> H <sub>19</sub> N <sub>3</sub> O <sub>2</sub>	66.0	7.0	15.4
(46)	Vinyl acetate	EtOH	50	1	60	150-151	EtOH	62.8	6.1	14.5	C <sub>15</sub> H <sub>17</sub> N <sub>3</sub> O <sub>3</sub>	62.7	5.9	14.6
(47)	<i>N</i> -Vinylpyrrolidin-2-one	EtOH	50	3	4	180-181	C <sub>6</sub> H <sub>12</sub>	65.2	6.6	17.7	C <sub>17</sub> H <sub>20</sub> N <sub>4</sub> O <sub>2</sub>	65.4	6.4	17.9
(48)	$\alpha$ -Chloroacrylonitrile	MeCN	15	5	41	151-152	EtOH	58.0	4.2	19.7	C <sub>14</sub> H <sub>13</sub> ClN <sub>4</sub> O	58.2	4.5	19.4
(49)	$\alpha$ -Methylacrylonitrile	MeCN	15	2	34	186-187	EtOH	67.5	6.1	20.6	C <sub>15</sub> H <sub>16</sub> N <sub>4</sub> O	67.2	6.0	20.9
(50)	Isopropenyl acetate <sup>b</sup>	EtOH	15	2	6	92-93	Et <sub>2</sub> O-EtOH	63.4	6.4	13.6	C <sub>16</sub> H <sub>19</sub> N <sub>3</sub> O <sub>3</sub>	63.8	6.3	14.0
(51)	Diethyl maleate	MeCN	15	2	40	100-101	EtOH	61.3	6.2	11.0	C <sub>19</sub> H <sub>23</sub> N <sub>3</sub> O <sub>5</sub>	61.1	6.1	11.3
(52)	Diethyl maleate	MeCN	15	2	45	90-91	EtOH	61.0	6.2	11.1	C <sub>19</sub> H <sub>23</sub> N <sub>3</sub> O <sub>5</sub>	61.1	6.1	11.3
(53)	Cyclohexene <sup>b</sup>	EtOH	15	1	8	162-163	EtOH-Et <sub>2</sub> O	71.8	7.5	14.4	C <sub>17</sub> H <sub>21</sub> N <sub>3</sub> O	72.1	7.4	14.8
(54)	Ethyl phenylpropiolate	Chlorobenzene	20	7	15	155-156	EtOH <sup>c</sup>	64.6	4.7	10.5	C <sub>21</sub> H <sub>17</sub> N <sub>3</sub> O <sub>5</sub>	64.5	4.3	10.7
(55)	Ethyl phenylpropiolate	Chlorobenzene	20	7	6	162-163	EtOH <sup>c</sup>	64.4	4.5	10.5	C <sub>21</sub> H <sub>17</sub> N <sub>3</sub> O <sub>5</sub>	64.5	4.3	10.7

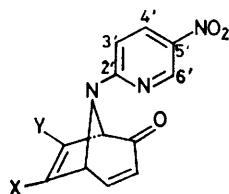
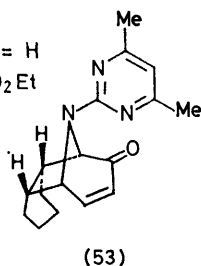
<sup>a</sup> Prisms unless otherwise stated. <sup>b</sup> Under sealed tube conditions at 80  $^\circ\text{C}$ . <sup>c</sup> Plates.

chloroacrylonitrile and  $\alpha$ -methylacrylonitrile formed the single 6-C isomers (48) and (49) respectively of unknown stereochemistry. Methyl vinyl ketone yielded the single 6-*exo*-isomer (40). The reaction of diethyl maleate with



- (40) X = COMe, W = Z = Y = H  
 (41) X = CN, W = Z = Y = H  
 (42) X = H; Y = CN; W = Z = H  
 (43) X = H; Y = CO<sub>2</sub>Me; W = Z = H  
 (44) X = CO<sub>2</sub>Me; W = Z = Y = H  
 (45) X = H; W = Z = H; Y = OEt  
 (46) X = H; W = Z = H; Y = OCOMe  
 (47) X = H; Y = ; W = Z = H

- (48) X, Y = Cl, CN; W = Z = H  
 (49) X, Y = Me, CN; W = Z = H  
 (50) X, Y = Me, OCOMe; W = Z = H  
 (51) X = H; Y = CO<sub>2</sub>Et; Z = CO<sub>2</sub>Et; W = H  
 (52) X = CO<sub>2</sub>Et; Z = H; Y = H; W = CO<sub>2</sub>Et



- (54) and (55) X, Y = Ph, CO<sub>2</sub>Et

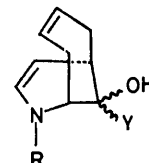
the pyrimidinyl betaine, generated *in situ* with triethylamine, yielded the two *trans*-isomers (51) and (52). FMO calculations<sup>9</sup> predict the formation of the 6-*endo*,7-*endo*-bis(ethoxycarbonyl) adduct. Presumably base-catalysed isomerism<sup>11</sup> of the initially formed kinetically controlled *cis*-product produced the less strained *trans*-isomers. Cyclohexene yielded a single 6,7-*endo*-adduct (53), the formation of which attests to the high reactivity of the pyrimidinyl betaine. The acetylenic dipolarophile, ethyl phenyl propiolate readily produced the adducts (54) and (55), whose structures could not be differentiated by spectral means.

**Adducts with Dienes.**—Butadiene reacted across the 2,4-positions of betaines (13a and b) to give the expected adducts (56a and b); we have previously reported<sup>12</sup> the analogous reaction with (13c). Bromination of (56a and b) occurred exclusively at the enamine group to yield (57a and b) as shown by the n.m.r. spectra (see below)

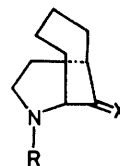
(Table 4). The pyrimidinyl adduct (56b) formed the oxime (58b), while the pyridyl adduct (56a) yielded the phenylhydrazone (59a). Reduction of (56b) with sodium borohydride gave the alcohol (60b) and with PhMgBr the alcohol (61b). Catalytic reduction of the adduct (56b) gave the saturated ketone (62b) which formed an oxime (63b) and was reduced to the alcohol (64b).



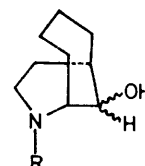
- (56) X = O, Y = H  
 (57) X = O, Y = Br  
 (58) X = NOH, Y = H  
 (59) X = NNHPh, Y = H



- (60) Y = H  
 (61) Y = Ph



- (62) X = O  
 (63) X = NOH



- (64)

- a; R = 5-Nitro-2-pyridyl  
 b; R = 4,6-Dimethylpyrimidin-2-yl

#### EXPERIMENTAL

The melting points were determined on a Reichert apparatus. Spectra were recorded with a Perkin-Elmer 257 grating infrared spectrophotometer, a Unicam ultraviolet spectrophotometer, a Hitachi-Perkin-Elmer RMU-6E mass spectrometer, and a Varian HA-100 n.m.r. spectrometer. Compounds were purified until they were observed as a single spot on t.l.c. (Kieselgel GF 254, type 60), and Kieselgel PF 254 was used for the preparative t.l.c. operations.

**9-(5-Nitro-2-pyridyl)-endo-5-oxa-9-azatricyclo[5.2.1.0<sup>4,8</sup>]-decan-2-one (10a).**—The dimer (15a) (1 g, 0.0046 mol betaine) in allyl alcohol (20 ml) was heated under reflux for 2 d and the reaction monitored by t.l.c. (CHCl<sub>3</sub>). Excess of allyl alcohol was removed *in vacuo*. The residue was chromatographed over aluminium oxide (BDH Alumina neutral, 60 g, toluene), followed by preparative t.l.c. (CHCl<sub>3</sub>). The *decanone* (10a) (0.60 g, 48%), was obtained as yellow prisms, m.p. 217–218 °C (1,2-dichloroethane) (Found: C, 56.6; H, 4.8; N, 15.0. C<sub>13</sub>H<sub>13</sub>N<sub>3</sub>O<sub>4</sub> requires C, 56.7; H, 4.7; N, 15.3%);  $\nu_{\max}$  (Nujol) 1720 cm<sup>-1</sup> (saturated CO);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 355 nm (log  $\epsilon$  368); *m/e* 275 (58%).

**9-(4,6-Dimethylpyrimidin-2-yl)-endo-5-oxa-9-azatricyclo[5.2.1.0<sup>4,8</sup>]-decan-2-one (10b).**—A solution of the dimer (16b) (1 g, 0.005 mol betaine) in allyl alcohol (20 ml) was heated under reflux for 2 d. The excess of allyl alcohol was removed *in vacuo*. The residue was washed with light petroleum (b.p. 40–60 °C) (*ca.* 10 ml), and purified by t.l.c. on silica gel (CHCl<sub>3</sub>) to give the *title compound* (10b) (1.1 g,

TABLE 4  
<sup>1</sup>H N.m.r. data (at 100 MHz) for butadiene adducts <sup>a</sup>

(a) Chemical shifts											
Proton	(56a) <sup>b</sup>	(56b) <sup>b</sup>	(57a) <sup>b</sup>	(57b) <sup>b</sup>	(58b) <sup>c</sup>	(59a) <sup>c</sup>	(60b) <sup>b</sup>	(61b) <sup>b</sup>	(62b) <sup>b</sup>	(63b) <sup>c</sup>	(64b) <sup>b</sup>
1	3.22 <sup>d</sup>	3.28 <sup>d</sup>	3.46 <sup>e</sup>	3.24 <sup>f</sup>	3.38 <sup>g</sup>	3.40 <sup>g</sup>	3.06 <sup>g</sup>	2.80— 3.10 <sup>d</sup>	2.80 <sup>d</sup>	3.30 <sup>d</sup>	3.18 <sup>g</sup>
2- <i>exo</i>	2.10 <sup>g</sup>	2.10 <sup>g</sup>	2.12 <sup>g</sup>	1.80— 3.00 <sup>d</sup>	2.10— 2.80 <sup>d</sup>	2.00— 3.00 <sup>d</sup>	2.70 <sup>d</sup>	2.80— 3.10 <sup>d</sup>	1.90 <sup>d</sup>	1.50— 1.90 <sup>d</sup>	1.40— 2.40 <sup>d</sup>
2- <i>endo</i>	3.10 <sup>d</sup>	3.28 <sup>d</sup>	3.04 <sup>g</sup>	1.80— 3.00 <sup>d</sup>	2.10— 2.80 <sup>d</sup>	2.00— 3.00 <sup>d</sup>	2.70 <sup>d</sup>	2.80— 3.10 <sup>d</sup>	1.90 <sup>d</sup>	1.50— 1.90 <sup>d</sup>	1.40— 2.40 <sup>d</sup>
3,4	5.72 <sup>e</sup>	5.72 <sup>e</sup>	5.86 <sup>e</sup>	5.68 <sup>e</sup>	5.64 <sup>h</sup>	5.50 <sup>h</sup>	5.48 <sup>d</sup>	5.56 <sup>i</sup>	1.58 <sup>d</sup>	1.50— 1.90 <sup>d</sup>	1.40— 2.40 <sup>d</sup>
5- <i>exo</i>	2.28 <sup>d</sup>	2.40 <sup>d</sup>	2.28 <sup>g</sup>	1.80— 3.00 <sup>d</sup>	2.10— 2.80 <sup>d</sup>	2.00— 3.00 <sup>d</sup>	2.70 <sup>d</sup>	2.80— 3.10 <sup>d</sup>	1.90 <sup>d</sup>	1.50— 1.90 <sup>d</sup>	1.40— 2.40 <sup>d</sup>
5- <i>endo</i>	2.40 <sup>g</sup>	2.40 <sup>d</sup>	2.84 <sup>g</sup>	1.80— 3.00 <sup>d</sup>	2.10— 2.80 <sup>d</sup>	2.00— 3.00 <sup>d</sup>	2.70 <sup>d</sup>	2.80— 3.10 <sup>d</sup>	1.90 <sup>d</sup>	1.50— 1.90 <sup>d</sup>	1.40— 2.40 <sup>d</sup>
6	5.05 <sup>j</sup>	5.19 <sup>j</sup>	5.06 <sup>e</sup>	5.06 <sup>f</sup>	5.00 <sup>d</sup>	5.14 <sup>g</sup>	4.80 <sup>g</sup>	5.24 <sup>g</sup>	5.30 <sup>g</sup>	5.05 <sup>g</sup>	5.20 <sup>g</sup>
8	7.16 <sup>i</sup>	7.72 <sup>i</sup>	7.58 <sup>h</sup>	7.98 <sup>h</sup>	6.49 <sup>j</sup>	7.10 <sup>d</sup>	7.67 <sup>i</sup>	7.65 <sup>d</sup>	4.42 <sup>f</sup>	2.50 <sup>g</sup>	4.50 <sup>k</sup>
9	4.87 <sup>j</sup>	4.62 <sup>j</sup>			7.28 <sup>j</sup>	4.86 <sup>j</sup>	4.62 <sup>j</sup>	4.45 <sup>j</sup>	3.66 <sup>k</sup>	1.50— 1.90 <sup>d</sup>	1.40— 2.40 <sup>d</sup>
10					5.00 <sup>d</sup>		4.43 <sup>e</sup>				4.06 <sup>e</sup>
3'	6.72 <sup>i</sup>		6.80 <sup>i</sup>			7.34 <sup>i</sup>					
4'	8.29 <sup>j</sup>		8.36 <sup>j</sup>			8.23 <sup>i</sup>					
5'		6.42 <sup>h</sup>		6.36 <sup>h</sup>	6.38 <sup>h</sup>		6.38 <sup>h</sup>	6.38 <sup>h</sup>	6.25 <sup>h</sup>	6.38 <sup>h</sup>	6.20 <sup>h</sup>
6'	9.10 <sup>i</sup>		9.12 <sup>i</sup>			8.96 <sup>i</sup>					
4',6' Me		2.28 <sup>h</sup>		2.20 <sup>h</sup>	2.20 <sup>h</sup>		2.29 <sup>h</sup>	2.30 <sup>h</sup>	2.24 <sup>h</sup>	2.20 <sup>h</sup>	2.25 <sup>h</sup>
Ph						7.05 <sup>d</sup>		7.20— 7.60 <sup>d</sup>			
NH						9.18 <sup>h</sup>					
NOH					10.39 10.80 <sup>h</sup>					10.21 <sup>h</sup> 10.46 <sup>h</sup>	
(b) Coupling constants (Hz)											
	(56a)	(56b)	(57a)	(57b)	(58b)	(59a)	(60b)	(61b)	(62b)	(63b)	(64b)
<i>J</i> (1,2- <i>exo</i> )			5	5	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>
<i>J</i> (1,2- <i>endo</i> )	5	5	5	5	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>
<i>J</i> (1,9)					4	6	6				
<i>J</i> (1,6)	2	2	2	2							
<i>J</i> (1,10)							5				6
<i>J</i> (2- <i>exo</i> ,2- <i>endo</i> )	16	16	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>
<i>J</i> (2- <i>exo</i> ,3)	3	5	5	5	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>
<i>J</i> (2- <i>endo</i> ,3)	3	5	5	5	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>
<i>J</i> (3,4)	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	7	6	<i>l</i>	<i>l</i>	<i>l</i>
<i>J</i> (4,5- <i>exo</i> )	3	5	5	5	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>
<i>J</i> (4,5- <i>endo</i> )	3	5	5	5	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>
<i>J</i> (5- <i>exo</i> ,5- <i>endo</i> )	14	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>
<i>J</i> (5- <i>exo</i> ,6)			5	5	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	5	<i>l</i>	<i>l</i>
<i>J</i> (5- <i>endo</i> ,6)	6	<i>l</i>	5	5	<i>l</i>	<i>l</i>	<i>l</i>	<i>l</i>	5	<i>l</i>	<i>l</i>
<i>J</i> (6,10)							5				6
<i>J</i> (8,9)	9	9			8	10	8	9	5, 7	<i>l</i>	
<i>J</i> (8A,8B)									14		14
<i>J</i> (3',4')	11		<i>l</i>	<i>l</i>		10					
<i>J</i> (4',6')	3		<i>l</i>	<i>l</i>		3					

<sup>a</sup> In p.p.m. relative to SiMe<sub>4</sub> as internal standard. <sup>b</sup> In CDCl<sub>3</sub>. <sup>c</sup> In (CD<sub>3</sub>)<sub>2</sub>SO. <sup>d</sup> Overlapped with other signals. <sup>e</sup> Triplet. <sup>f</sup> Double triplet. <sup>g</sup> Multiplet. <sup>h</sup> Singlet. <sup>i</sup> Doublet. <sup>j</sup> Double doublet. <sup>k</sup> Quartet of doublets. <sup>l</sup> Not measurable owing to signal overlap.

85%) as prisms, m.p. 120—121 °C (EtOH) (Found: C, 64.4; H, 6.4; N, 15.9. C<sub>14</sub>H<sub>17</sub>N<sub>3</sub>O<sub>2</sub> requires C, 64.9; H, 6.6; N, 16.2%);  $\nu_{\max}$  (Nujol) 1 725 cm<sup>-1</sup> (saturated CO);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 293 (log  $\epsilon$  3.50) and 250 nm (4.10); *m/e* 259 (74%). The reaction also gave 8-(4,6-dimethylpyrimidin-2-yl)-6-*exo*-(hydroxymethyl)-8-azabicyclo[3.2.1]oct-3-*en*-2-*one* (22b) (0.2 g, 14%) as a yellow gum which resisted recrystallisation (Found: C, 64.1; H, 7.0; N, 16.4. C<sub>14</sub>H<sub>17</sub>N<sub>3</sub>O<sub>2</sub> requires C, 64.9; H, 6.6; N, 16.2%);  $\nu_{\max}$  (Nujol) 1 680 cm<sup>-1</sup> ( $\alpha\beta$ -unsaturated ketone CO);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 251 (log  $\epsilon$  6.10) and 295 nm (5.49); *m/e* 259 (65%).

9-(4,6-Dimethoxy-*s*-triazin-2-yl)-endo-5-*oxa*-9-azatricyclo-[5.2.1.0<sup>4,8</sup>]decan-2-*one* (10c).—A solution of the dimer (15c) (1 g, 0.004 2 mol betaine) in allyl alcohol (25 ml) was heated under reflux for 2 d and the reaction monitored by t.l.c. (CHCl<sub>3</sub>). The reaction mixture was then evaporated *in vacuo* and washed with light petroleum (b.p. 40—60 °C) (3 × 5 ml). The residue was purified by preparative t.l.c.

(CHCl<sub>3</sub>) to yield the *title adduct* (10c) (122 mg, 10%) as needles, m.p. 120—121 °C (1,2-dichloroethane) (Found: C, 53.3; H, 5.3; N, 19.0. C<sub>13</sub>H<sub>16</sub>N<sub>4</sub>O<sub>4</sub> requires C, 53.4; H, 5.5; N, 19.2%);  $\nu_{\max}$  (Nujol) 1 730 (saturated CO), 1 590, 1 570, and 1 550 cm<sup>-1</sup> (C-N);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 247 nm (log  $\epsilon$  3.88); *m/e* 292 (43%).

5-Benzenesulphonyl-9-(5-nitro-2-pyridyl)-5,9-diazatricyclo-[5.2.1.0<sup>4,8</sup>]decan-2-*one* (11a).—A solution of the dimer (15a) (0.5 g, 0.002 3 mol betaine) and allylbenzenesulphonamide (2.8 g, 0.014 2 mol) in toluene (5 ml) was heated under reflux for 2 d and the reaction monitored by t.l.c. (CHCl<sub>3</sub>). The solvent was evaporated off *in vacuo*, and the brown residue, after treatment with light petroleum (b.p. 40—60 °C) (15 ml), was purified by preparative t.l.c. (CHCl<sub>3</sub>) to give the pure *compound* (11a) (0.210 g, 22%) as yellow prisms, m.p. 198—199 °C (1,2-dichloroethane) (Found: C, 54.8; H, 4.4; N, 13.4; S, 7.6. C<sub>19</sub>H<sub>18</sub>N<sub>4</sub>O<sub>5</sub>S requires C, 55.1; H, 4.3; N, 13.5; S, 7.7%);  $\nu_{\max}$  (Nujol) 1 730 cm<sup>-1</sup>

(saturated CO);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 245 (log  $\epsilon$  3.74) and 352 nm (4.20);  $m/e$  414 (10%). The reaction also gave 6-endo-(*N*-benzenesulphonylaminoethyl)-8-(5-nitro-2-pyridyl)-8-azabicyclo[3.2.1]oct-3-en-2-one (18a) (0.19 g, 20%) as yellow prisms, m.p. 207—208 °C (dichloroethane) (Found: C, 55.3; H, 4.5; N, 13.5; S, 7.7. C<sub>19</sub>H<sub>18</sub>N<sub>4</sub>O<sub>5</sub>S requires C, 55.1; H, 4.3; N, 13.5; S, 7.7%);  $\nu_{\max}$  (Nujol) 1 675 cm<sup>-1</sup> (unsaturated CO);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 245 (log  $\epsilon$  3.65) and 348 nm (4.04);  $m/e$  414 (38%).

5-Benzenesulphonyl-9-(4,6-dimethylpyrimidin-2-yl)-5,9-diazatricyclo[5.2.1.0<sup>4,8</sup>]decan-2-one (11b).—A solution of the dimer (16b) (0.5 g, 0.002 4 mol betaine) and allylbenzenesulphonamide (3.7 g, 0.018 7 mol) was refluxed in toluene (5 ml) for 2 d with monitoring by t.l.c. (CHCl<sub>3</sub>). The toluene was removed *in vacuo*, and the semi-solid residue was treated with light petroleum (b.p. 40—60 °C) (20 ml) to remove the excess of dipolarophile. The crude product was purified by preparative t.l.c. (CHCl<sub>3</sub>) to yield compound (11b) (0.65 g, 68%) as prisms, m.p. 212—213 °C (1,2-dichloroethane) (Found: C, 60.0; H, 5.9; N, 14.2; S, 8.2. C<sub>20</sub>H<sub>22</sub>N<sub>4</sub>O<sub>3</sub>S requires C, 60.3; H, 5.5; N, 14.1; S, 8.0%);  $\nu_{\max}$  (Nujol) 1 720 cm<sup>-1</sup> (saturated CO);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 248 (log  $\epsilon$  4.21) and 290 nm (3.66);  $m/e$  398 (25%).

3-Benzylideno-9-(4,6-dimethylpyrimidin-2-yl)-5-oxa-9-azatricyclo[5.2.1.0<sup>4,8</sup>]decan-2-one (12b).—A solution of benzaldehyde (0.3 g, 0.002 8 mol) in EtOH (5 ml) was added to a solution of compound (10b) (0.5 g, 0.001 9 mol) and KOH (0.4 g, 0.007 mol) in a mixture of EtOH (20 ml) and H<sub>2</sub>O (5 ml). After stirring at room temperature for 2 h, the yellow precipitate was filtered off and purified by preparative t.l.c. (CHCl<sub>3</sub>) to give the derivative (12b) (0.35 g, 53%), as yellow prisms, m.p. 135—136 °C (Et<sub>2</sub>O) (Found: C, 72.3; H, 6.1; N, 12.0. C<sub>21</sub>H<sub>21</sub>N<sub>3</sub>O<sub>2</sub> requires C, 72.6; H, 6.1; N, 12.1%);  $\nu_{\max}$  (Nujol) 1 695 cm<sup>-1</sup> (unsaturated CO);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 250 (log  $\epsilon$  4.21) and 295 nm (4.25);  $m/e$  347 (100%).

6-endo-(Acetoxyethyl)-8-(5-nitro-2-pyridyl)-8-azabicyclo[3.2.1]oct-3-en-2-one (17a).—The dimer (15a) (1 g, 0.002 3 mol) in allyl acetate (25 ml) was heated under reflux for 2 d and the reaction monitored by t.l.c. (CHCl<sub>3</sub>). Excess of dipolarophile was removed *in vacuo*, and the residue purified by preparative t.l.c. [EtOAc-CHCl<sub>3</sub> (1:1)]. The title compound (17a) (0.394 g, 27%) was isolated as yellow prisms, m.p. 145—147 °C [EtOH-cyclohexane (50:1)] (Found: C, 56.6; H, 4.6; N, 13.3. C<sub>15</sub>H<sub>15</sub>N<sub>3</sub>O<sub>5</sub> requires C, 56.8; H, 4.7; N, 13.2%);  $\nu_{\max}$  (Nujol) 1 740 (ester CO) and 1 670 cm<sup>-1</sup> (unsaturated CO);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 350 nm (log  $\epsilon$  3.92);  $m/e$  317 (46%).

*t*-Butyl 8-(3-*p*-Bromophenyl-3-oxopropen-1-yl)-2-oxo-8-azabicyclo[3.2.1]oct-3-ene-6-exo-carboxylate (23d) and -6-endo-carboxylate (19d).—A solution of the betaine (13d) (0.5 g, 0.001 6 mol) and *t*-butyl acrylate (3 ml) in MeCN (20 ml) was refluxed gently for 3 d. The reaction was monitored by t.l.c. [EtOAc-light petroleum (1:1)]. The mixture was concentrated *in vacuo* and the pure isomers were separated by preparative t.l.c. [EtOAc-light petroleum (b.p. 40—60 °C) (1:1)]. The endo-isomer (19d) (0.11 g, 16%) was isolated as pinkish prisms, m.p. 190—191 °C (Et<sub>2</sub>O) (Found: C, 58.7; H, 5.4; N, 3.1; Br, 18.2. C<sub>21</sub>H<sub>22</sub>BrNO<sub>4</sub> requires C, 58.3; H, 5.1; N, 3.2; Br, 18.5%);  $\nu_{\max}$  (Nujol) 1 730 (ester CO) and 1 690 cm<sup>-1</sup> (unsaturated CO);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 265 (log  $\epsilon$  4.27) and 334 nm (4.43);  $m/e$  432 (11%). The exo-isomer (23d) (115 mg, 17%) was isolated as pinkish prisms, m.p. 165—166 °C (Et<sub>2</sub>O) (Found: C, 58.6; H, 5.4; N, 3.2; Br, 18.6%);  $\nu_{\max}$  (Nujol) 1 720 (ester CO) and

1 690 cm<sup>-1</sup> (unsaturated CO);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 265 (log  $\epsilon$  4.23) and 334 nm (4.41);  $m/e$  432 (8%).

8-(4,6-Dimethylpyrimidin-2-yl)-6-endo-(4-pyridyl)-8-azabicyclo[3.2.1]oct-3-en-2-one (20b).—A solution of the salt (14b) (1 g, 0.004 2 mol), hydroquinone (200 mg), 4-vinylpyridine (5 ml), and Et<sub>3</sub>N (4 ml) in MeCN (15 ml) was heated under reflux at 100 °C for 4 d. The mixture was concentrated *in vacuo* and filtered to remove precipitated Et<sub>3</sub>N·HCl. The filtrate was evaporated to dryness *in vacuo* and the residue purified by preparative t.l.c. (CHCl<sub>3</sub>). The cycloadduct (20b) (0.15 g, 12%) was isolated as prisms, m.p. 150 °C [EtOH-cyclohexane (1:1)] (Found: C, 70.4; H, 6.0; N, 18.2. C<sub>18</sub>H<sub>18</sub>N<sub>4</sub>O requires C, 70.6; H, 5.8; N, 18.3%);  $\nu_{\max}$  (Nujol) 1 690 cm<sup>-1</sup> (unsaturated CO);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 254 (log  $\epsilon$  4.20) and 290 nm (3.59);  $m/e$  306 (47%).

8-(4,6-Dimethylpyrimidin-2-yl)-6-endo-(2-pyridyl)-8-azabicyclo[3.2.1]oct-3-en-2-one (21b).—A mixture of the cycloadduct (36) (100 mg, 0.000 2 mol) in H<sub>2</sub>O (5 ml) and KOH solution (200 mg, 0.003 5 mol in 5 ml water) was stirred at room temperature for 30 min. The precipitate was collected and washed with H<sub>2</sub>O to give the product (21b) (55 mg, 90%) as prisms, m.p. 157—158 °C [EtOH-Et<sub>2</sub>O (1:10)] (Found: C, 70.3; H, 6.0; N, 18.0. C<sub>18</sub>H<sub>18</sub>N<sub>4</sub>O requires C, 70.6; H, 5.9; N, 18.3%);  $\nu_{\max}$  (Nujol) 1 685 cm<sup>-1</sup> (unsaturated CO);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 256 (log  $\epsilon$  4.20) and 293 nm (3.56);  $m/e$  306 (55%).

Quaternization of Compound (21b).—A solution of (21b) (0.7 g, 0.002 2 mol) in EtOAc (50 ml) and MeI (25 ml) was refluxed for 7 d and the reaction monitored by t.l.c. (CHCl<sub>3</sub>). When the starting material was consumed, the reaction mixture was evaporated *in vacuo* to give the methiodide (38) (0.6 g, 60%) as yellowish prisms, m.p. 172—173 °C (MeCN) (Found: C, 50.6; H, 4.8; N, 12.7; I, 28.2. C<sub>19</sub>H<sub>21</sub>IN<sub>4</sub>O requires C, 50.9; H, 4.7; N, 12.5; I, 28.3%);  $\nu_{\max}$  (Nujol) 1 700 cm<sup>-1</sup> (saturated CO);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 248 nm (log  $\epsilon$  4.31).

8-Allyloxy-3,11-bis(4,6-dimethoxy-s-triazin-2-yl)-3,11-diazatricyclo[5.3.1.1<sup>3,6</sup>]dodecan-4-ene-10,12-dione (24c).—Et<sub>3</sub>N (1 ml) was added dropwise during 20 min to a well-stirred solution of salt (13c) (1 g, 0.003 7 mol) and allyl alcohol (1.6 g, 0.027 5 mol) in 1,2-dichloroethane (20 ml) at room temperature. After 2 h, the violet colour faded. The precipitated Et<sub>3</sub>N·HCl was filtered off and the filtrate was concentrated *in vacuo*. The residue was purified by preparative t.l.c. [on silica gel (CHCl<sub>3</sub>)] to yield the cycloadduct (24c) (0.486 g, 25%), as prisms, m.p. 185—186 °C [CH<sub>2</sub>Cl<sub>2</sub>-1,2-dichloroethane (1:2)] (Found: C, 52.0; H, 4.9; N, 21.2. C<sub>23</sub>H<sub>26</sub>N<sub>8</sub>O<sub>7</sub> requires C, 52.5; H, 4.9; N, 21.3%);  $\nu_{\max}$  (Nujol) 1 730 (saturated CO) and 1 550 cm<sup>-1</sup> (C-N);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 270 nm (log  $\epsilon$  4.2);  $m/e$  526 (12%).

9-(4,6-Dimethylpyrimidin-2-yl)-5-oxa-9-azatricyclo[5.2.1.0<sup>4,8</sup>]decan-2-one phenylhydrazine (25).—Phenylhydrazine hydrochloride (0.7 g, 0.005 mol) and NaOAc (1 g, 0.001 2 mol) in H<sub>2</sub>O (10 ml) were added to a well-stirred solution of the cycloadduct (10b) (0.4 g, 0.001 5 mol) in water (10 ml). After 2 h, the reaction mixture was diluted with H<sub>2</sub>O (50 ml) and the precipitate collected by filtration. The crude product was washed with H<sub>2</sub>O and crystallised from EtOH to give the phenylhydrazine (25) (0.5 g, 96%) as yellow prisms, m.p. 227—228 °C (Found: C, 68.4; H, 6.7; N, 19.7. C<sub>20</sub>H<sub>23</sub>N<sub>5</sub>O requires C, 68.7; H, 6.6; N, 20.0%);  $\nu_{\max}$  (Nujol) 3 270 (NH) and 1 600 cm<sup>-1</sup> (C-N);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 348 nm (log  $\epsilon$  4.02);  $m/e$  349 (100%).

9-(4,6-Dimethylpyrimidin-2-yl)-2-morpholino-5-oxa-9-azatricyclo[5.2.1.0<sup>4,8</sup>]decan-2-ene (26).—A mixture of the cyclo-



adduct (10b) (1 g, 0.0038 mol), toluene-*p*-sulphonic acid (0.65 g, 0.003 mol), morpholine (3.5 g, 0.04 mol), and molecular sieves (4 Å, 4 g) in toluene (15 ml) was heated under reflux at 100 °C. The reaction was monitored by t.l.c. (CHCl<sub>3</sub>). When all the starting cycloadduct was consumed, the reaction was cooled to room temperature and filtered. The filtrate was evaporated to dryness and the residue was purified by preparative t.l.c. (CHCl<sub>3</sub>) to give the adduct (26) (200 mg, 16%) as prisms, m.p. 132–133 °C (cyclohexane) (Found: C, 65.8; H, 7.5; N, 16.8. C<sub>18</sub>H<sub>24</sub>N<sub>4</sub>O<sub>2</sub> requires C, 65.9; H, 7.3; N, 16.8%);  $\nu_{\max}$  (Nujol) 1620 cm<sup>-1</sup> (C=C-N);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 250 nm (log  $\epsilon$  4.02); *m/e* 328 (100%).

9-(4,6-Dimethylpyrimidin-2-yl)-5-oxa-9-azatricyclo[5.2.1.0<sup>4,8</sup>]decan-2-ol (27).—A solution of the cycloadduct (10b) (0.5 g, 0.0019 mol) in MeOH (25 ml) was treated with a solution of NaBH<sub>4</sub> (50 mg, in 2 ml of 0.2N NaOH) and stirred at room temperature for 1 h. The mixture was evaporated to dryness and extracted with Et<sub>2</sub>O (4 × 10 ml). The extract was purified by preparative t.l.c. (CHCl<sub>3</sub>) to yield the alcohol (27) (0.4 g, 81%) as prisms, m.p. 90–91 °C (Et<sub>2</sub>O) (Found: C, 64.2; H, 7.6; N, 15.9. C<sub>14</sub>H<sub>19</sub>N<sub>3</sub>O<sub>2</sub> requires C, 64.4; H, 7.3; N, 16.1%);  $\nu_{\max}$  (Nujol) 3380 cm<sup>-1</sup> (OH);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 256 (log  $\epsilon$  4.08) and 306 nm (3.52); *m/e* 261 (57%).

9-(5-Bromo-4,6-dimethylpyrimidin-2-yl)-5-oxa-9-azatricyclo[5.2.1.0<sup>4,8</sup>]decan-2-one (28).—A solution of bromine (182 mg, 0.0011 mol) in AcOH (5 ml) was added dropwise to a well-stirred solution of compound (10b) (0.45 g, 0.0017 mol) in AcOH (7.5 ml). After 1 h the solution was basified with NH<sub>4</sub>OH and extracted with CHCl<sub>3</sub>. The combined extracts when concentrated *in vacuo* gave the crude product which was purified by preparative t.l.c. (CHCl<sub>3</sub>). The desired product (28) (0.3 g, 76%) was obtained as prisms, m.p. 95–96 °C [EtOH–Et<sub>2</sub>O (1:10)] (Found: C, 49.7; H, 4.9; N, 12.3; Br, 23.5. C<sub>14</sub>H<sub>16</sub>BrN<sub>3</sub>O<sub>2</sub> requires C, 49.7; H, 4.7; N, 12.4; Br, 23.6%);  $\nu_{\max}$  (Nujol) 1725 cm<sup>-1</sup> (saturated CO);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 257 nm (log  $\epsilon$  4.38); *m/e* 337 (22%) and 339 (22%).

9-(5-Bromo-4,6-dimethylpyrimidin-2-yl)-3,3-dibromo-5-oxa-9-azatricyclo[5.2.1.0<sup>4,8</sup>]decan-2-one (29).—A solution of bromine (0.23 g, 0.0014 mol) in pyridine (5 ml) was added dropwise during 1 h to a well-stirred solution of the cycloadduct (10b) (0.5 g, 0.0019 mol) in pyridine (10 ml). The reaction mixture was evaporated to dryness *in vacuo* at 100 °C, and the residue obtained was treated with light petroleum (25 ml) (b.p. 40–60 °C) to remove unreacted bromine. The crude mixture was purified by preparative t.l.c. (CHCl<sub>3</sub>) to give the bromo-derivative (29) (0.35 g, 37%) as prisms, m.p. 206–207 °C (1,2-dichloroethane) (Found: C, 33.7; H, 3.0; N, 8.3; Br, 48.4. C<sub>14</sub>H<sub>14</sub>Br<sub>3</sub>N<sub>3</sub>O<sub>2</sub> requires C, 33.9; H, 2.8; N, 8.5; Br, 48.4%);  $\nu_{\max}$  (Nujol) 1740 cm<sup>-1</sup> (saturated CO);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 253 nm (log  $\epsilon$  4.22).

3,3-Dibromo-9-(5-nitro-2-pyridyl)-5-oxa-9-azatricyclo[5.2.1.0<sup>4,8</sup>]decan-2-one (30) and 9-(3-Bromo-5-nitro-2-pyridyl)-3,3-dibromo-5-oxa-9-azatricyclo[5.2.1.0<sup>4,8</sup>]decan-2-one (31).—Bromine was added dropwise to a solution of the cycloadduct (10a) (0.6 g, 0.0021 mol) in pyridine (40 ml) until the colour of bromine remained. After stirring at room temperature for 1 h, pyridine was removed *in vacuo* at 100 °C, and the residue obtained was extracted with portions of 1,2-dichloroethane (2 × 10 ml). The crude extracts were purified by preparative t.l.c. (CHCl<sub>3</sub>) to give the dibromo-derivative (30) (0.45 g, 51%), yellow prisms, m.p. 225–226 °C (1,2-dichloroethane) (Found: C, 34.1;

H, 2.7; N, 9.8; Br, 38.0. C<sub>12</sub>H<sub>11</sub>Br<sub>2</sub>N<sub>3</sub>O<sub>4</sub> requires C, 34.2; H, 2.6; N, 10.0; Br, 38.0%);  $\nu_{\max}$  (Nujol) 1750 cm<sup>-1</sup> (saturated CO);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 355 (log  $\epsilon$  4.31) nm. The compound (31) (0.117 g, 11%) was recrystallised from 1,2-dichloroethane as yellow prisms, m.p. 215–216 °C (Found: C, 28.7; H, 2.3; N, 8.3; Br, 47.9. C<sub>12</sub>H<sub>10</sub>Br<sub>3</sub>N<sub>3</sub>O<sub>4</sub> requires C, 28.8; H, 2.0; N, 8.4; Br, 48.0%);  $\nu_{\max}$  (Nujol) 1740 cm<sup>-1</sup> (saturated CO);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 355 nm (log  $\epsilon$  4.11).

Preparation of Compound (32).—A solution of the cycloadduct (10b) (0.5 g, 0.0019 mol), benzofuroxan (0.6 g, 0.0045 mol), and KOH (0.56 g, 0.01 mol) in EtOH (40 ml) was heated under reflux. When all the starting cycloadduct had been consumed (t.l.c., CHCl<sub>3</sub>), the solution was cooled and filtered. The filtrate was concentrated *in vacuo* and the residue was purified by preparative t.l.c. (CHCl<sub>3</sub>). The pure compound (32) (150 mg, 20%) was obtained as orange prisms, m.p. 215–216 °C [EtOH–Et<sub>2</sub>O (1:1)] (Found: C, 63.5; H, 5.3; N, 18.3. C<sub>20</sub>H<sub>19</sub>N<sub>5</sub>O<sub>3</sub> requires C, 63.7; H, 5.0; N, 18.6%);  $\nu_{\max}$  (Nujol) 1570 and 1370 cm<sup>-1</sup> (N–O);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 248 (log  $\epsilon$  3.82) and 270 nm (3.87); *m/e* 377 (8%).

Preparation of Compound (33).—Potassium borohydride (0.5 g, 0.006 mol) was added in small portions to a stirred solution of compound (32) (50 mg, 0.0001 mol) and KOH (5 ml of 10% solution) in MeOH (10 ml). After 1 h, the reaction mixture was gently refluxed until no more colour fading was observed (orange to pale yellow). The reaction mixture was neutralised (glacial AcOH) and evaporated to dryness *in vacuo*. The residue was diluted with H<sub>2</sub>O and extracted with CHCl<sub>3</sub> (2 × 10 ml). The CHCl<sub>3</sub> extracts were purified by preparative t.l.c. (CHCl<sub>3</sub>) to give compound (33) (25 mg, 69%) as yellow prisms, m.p. 215–216 °C (Et<sub>2</sub>O) (Found: C, 66.6; H, 5.3; N, 19.6. C<sub>20</sub>H<sub>19</sub>N<sub>5</sub>O<sub>2</sub> requires C, 66.5; H, 5.3; N, 19.4%);  $\nu_{\max}$  (Nujol) 1580 and 1360 (N–O) cm<sup>-1</sup>;  $\lambda_{\max}$  (CHCl<sub>3</sub>) 253 nm (log  $\epsilon$  4.39); *m/e* 361 (68%).

10-Allylamino-3,11-bis(4,6-dimethylpyrimidin-2-yl)-3,11-diazatricyclo[5.3.1.1<sup>2,8</sup>]dodecan-4-ene-8,12-dione (34).—A solution of the dimer (16b) (0.5 g, 0.0012 mol), allylamine (15 ml), and toluene (10 ml) was heated under reflux at 100 °C for 2 d. The course of the reaction was monitored by t.l.c. (EtOAc). The reaction mixture was evaporated to dryness *in vacuo* and the residue purified by preparative t.l.c. on silica gel [EtOAc–CHCl<sub>3</sub> (1:1)]. The title compound (34) (0.247 g, 22%), was isolated as yellowish prisms, m.p. 150 °C [EtOH–Et<sub>2</sub>O (2:1)] (Found: C, 65.5; H, 6.5; N, 21.0. C<sub>25</sub>H<sub>29</sub>N<sub>7</sub>O<sub>2</sub> requires C, 65.4; H, 6.3; N, 21.4%);  $\nu_{\max}$  (Nujol) 1720 (saturated CO) and 1640 cm<sup>-1</sup> (C=C–N);  $\lambda_{\max}$  (CHCl<sub>3</sub>) 250 (log  $\epsilon$  4.33) and 278 nm (4.29); *m/e* 459 (29%).

9-(5-Nitro-2-pyridyl)-5-oxa-9-azatricyclo[5.2.1.0<sup>4,8</sup>]decane-2,6-dione (35a).—A solution of the dimer (15a) (0.8 g, 0.0038 mol betaine) and triethylammonium acrylate (2 g) in MeCN–H<sub>2</sub>O (1:1) was refluxed for 7 d. The precipitate was collected and washed several times with MeCN (4 × 5 ml). The dark brown solid was recrystallised from MeCN–H<sub>2</sub>O to give compound (35a) as brown prisms (0.48 g, 44%), m.p. 265–266 °C (Found: C, 53.7; H, 3.6; N, 14.3. C<sub>13</sub>H<sub>11</sub>N<sub>3</sub>O<sub>5</sub> requires C, 54.0; H, 3.8; N, 14.5%);  $\nu_{\max}$  (Nujol) 1770 (lactone CO) and 1720 cm<sup>-1</sup> (saturated CO); *m/e* 289 (15%).

9-(4,6-Dimethylpyrimidin-2-yl)-5-oxa-9-azatricyclo[5.2.1.0<sup>4,8</sup>]decane-2,6-dione (35b).—A solution of the salt (13b) (1 g, 0.0042 mol), triethylammonium acrylate (2 g), and Et<sub>3</sub>N (10 ml) in MeCN (15 ml) was heated under reflux

for 2 d. The course of the reaction was followed by t.l.c. ( $\text{CHCl}_3$ ). The reaction mixture was evaporated *in vacuo* and the residue treated with  $\text{CH}_2\text{Cl}_2$ . The precipitate was discarded and the filtrate was concentrated *in vacuo*. The concentrate was purified by preparative t.l.c. ( $\text{CHCl}_3$ ) to give the *adduct* (35b) (0.6 g, 52%) as prisms, m.p. 136—137 °C ( $\text{Et}_2\text{O}$ ) (Found: C, 61.4; H, 5.5; N, 15.0.  $\text{C}_{14}\text{H}_{15}\text{N}_3\text{O}_3$  requires C, 61.5; H, 5.5; N, 15.4%);  $\nu_{\text{max}}$  (Nujol) 1780 (lactone CO) and 1720  $\text{cm}^{-1}$  (saturated CO);  $\lambda_{\text{max}}$  ( $\text{CHCl}_3$ ) 247 (log  $\epsilon$  4.31) and 287 nm (3.67);  $m/e$  273 (45%).

**Preparation of Cycloadducts (36) and (37).**—A solution of the salt (13b) (0.9 g, 0.0037 mol), hydroquinone (200 mg), water (1 ml), 2-vinylpyridine (5 ml), and  $\text{Et}_3\text{N}$  (4 ml), in MeCN (20 ml) was heated under reflux at 100 °C for 3 d. The course of the reaction was monitored by t.l.c. ( $\text{CHCl}_3$ ). The reaction was cooled and the  $\text{Et}_3\text{N}\cdot\text{HCl}$  removed. The filtrate was evaporated *in vacuo* and the brown gum was triturated with 1,2-dichloroethane (15 ml) and cooled (0 °C). The *cycloadduct* (36) was precipitated as prisms (0.7 g, 55%), m.p. 235—236 °C [1,2-dichloroethane–EtOH (10:1)] (Found: C, 62.8; H, 5.7; N, 16.1; Cl, 10.5.  $\text{C}_{18}\text{H}_{19}\text{ClN}_4\text{O}$  requires C, 63.1; H, 5.5; N, 16.4; Cl, 10.4%);  $\nu_{\text{max}}$  (Nujol) 1720  $\text{cm}^{-1}$  (saturated CO);  $\lambda_{\text{max}}$  ( $\text{CHCl}_3$ ) 246 (log  $\epsilon$  4.27) and 270 nm (log  $\epsilon$  4.11);  $m/e$  342.5 (37%). The filtrate was evaporated to dryness and chromatographed (preparative t.l.c.,  $\text{CHCl}_3$ ) to give 8-(4,6-dimethylpyrimidin-2-yl)-6-exo-(2-pyridyl)-8-azabicyclo[3.2.1]oct-3-en-2-one (37) (0.1 g, 9%) as yellowish prisms, m.p. 105—106 °C [ $\text{Et}_2\text{O}$ –EtOH (10:1)] (Found: C, 70.4; H, 6.0; N, 18.2.  $\text{C}_{18}\text{H}_{18}\text{N}_4\text{O}$  requires C, 70.6; H, 5.9; N, 18.3%);  $\nu_{\text{max}}$  (Nujol) 1680  $\text{cm}^{-1}$  (unsaturated CO);  $\lambda_{\text{max}}$  ( $\text{CHCl}_3$ ) 250 (log  $\epsilon$  4.28) and 293 nm (3.63);  $m/e$  306 (40%).

**7-(5-Nitro-2-pyridyl)-7-azabicyclo[4.3.1]deca-3,8-dien-10-one (56a).**—A solution of the dimer (15a) (0.7 g, 0.0016 mol betaine), hydroquinone (200 mg), and liquid butadiene (10 ml) in 1,2-dichloroethane (10 ml) was heated in a Carius tube for 4 d at 80 °C. The solvent was removed *in vacuo* and the residue extracted with benzene. The extract was purified by preparative t.l.c. ( $\text{CHCl}_3$ ) to give the *adduct* (56a) (0.36 g, 83%) as yellow needles, m.p. 139—140 °C (EtOH) (Found: C, 61.8; H, 4.8; N, 15.4.  $\text{C}_{14}\text{H}_{13}\text{N}_3\text{O}_3$  requires C, 61.9; H, 4.8; N, 15.5%);  $\nu_{\text{max}}$  (Nujol) 1733  $\text{cm}^{-1}$  (saturated CO);  $\lambda_{\text{max}}$  ( $\text{CHCl}_3$ ) 386 nm (log  $\epsilon$  4.52);  $m/e$  271 (40%).

**7-(4,6-Dimethylpyrimidin-2-yl)-7-azabicyclo[4.3.1]deca-3,8-dien-10-one (56b).**—A solution of the dimer (16b) (2.5 g, 0.006 mol betaine), hydroquinone (200 mg), and liquid butadiene (10 ml) in 1,2-dichloroethane (10 ml) was heated in a Carius tube for 3 d at 80 °C. The solvent was evaporated off *in vacuo* and the residue extracted with benzene. The extract was purified by preparative t.l.c. ( $\text{CHCl}_3$ ) to give the *adduct* (56b) (0.94 g, 62%), as prisms, m.p. 100—101 °C (EtOH) (Found: C, 70.5; H, 6.7; N, 16.4.  $\text{C}_{15}\text{H}_{17}\text{N}_3\text{O}$  requires C, 70.6; H, 6.7; N, 16.4%);  $\nu_{\text{max}}$  (Nujol) 1725  $\text{cm}^{-1}$  (saturated CO);  $\lambda_{\text{max}}$  ( $\text{CHCl}_3$ ) 277 nm (log  $\epsilon$  4.50);  $m/e$  255 (33%).

**9-Bromo-7-(5-nitro-2-pyridyl)-7-azabicyclo[4.3.1]deca-3,8-dien-10-one (57a).**—Bromine (0.05 g, 0.0003 mol) in  $\text{CH}_2\text{Cl}_2$  (5 ml) was added dropwise during 20 min to a well-stirred solution of the *cycloadduct* (56a) (160 mg, 0.0006 mol) in a mixture of  $\text{CH}_2\text{Cl}_2$  (5 ml) and  $\text{CCl}_4$  (10 ml) at 0—5 °C. The precipitate was collected and purified by preparative t.l.c. ( $\text{CHCl}_3$ ) to give the *bromo-derivative* (57a) (0.189 g, 91%) as yellow prisms, m.p. 143—144 °C [ $\text{EtOH}\text{--}\text{CH}_2\text{Cl}_2$  (1:1)]

(Found: C, 47.7; H, 3.6; N, 11.7; Br, 23.2.  $\text{C}_{14}\text{H}_{12}\text{BrN}_3\text{O}_3$  requires C, 48.0; H, 3.5; N, 12.0; Br, 22.8%);  $\nu_{\text{max}}$  (Nujol) 1725  $\text{cm}^{-1}$  (saturated CO);  $\lambda_{\text{max}}$  ( $\text{CHCl}_3$ ) 350 nm (log  $\epsilon$  3.92);  $m/e$  349 (42%) and 351 (42%).

**9-Bromo-7-(4,6-dimethylpyrimidin-2-yl)-7-azabicyclo[4.3.1]deca-3,8-dien-10-one (57b).**—Bromine was added dropwise to a stirred solution of the *cycloadduct* (56b) (0.5 g, 0.0019 mol) in  $\text{CCl}_4$  (25 ml) at 0 °C until the bromine was no longer consumed (30 min). After 30 min, the yellow precipitate was filtered, washed with  $\text{CCl}_4$ , and purified by preparative t.l.c. ( $\text{CHCl}_3$ ) to give the *bromo-derivative* (57b) (0.3 g, 47%), as prisms, m.p. 150—151 °C (EtOH) (Found: C, 53.5; H, 5.0; N, 12.3; Br, 23.6.  $\text{C}_{15}\text{H}_{16}\text{BrN}_3\text{O}$  requires C, 53.7; H, 4.8; N, 12.5; Br, 24.0%);  $\nu_{\text{max}}$  (Nujol) 1725  $\text{cm}^{-1}$  (saturated CO);  $\lambda_{\text{max}}$  ( $\text{CHCl}_3$ ) 287 nm (log  $\epsilon$  4.24);  $m/e$  333 (21%) and 335 (21%).

**7-(4,6-Dimethylpyrimidin-2-yl)-7-azabicyclo[4.3.1]deca-3,8-dien-10-one Oxime (58b).**—A solution of the *cycloadduct* (56b) (0.5 g, 0.0019 mol) in EtOH (75 ml) was mixed with hydroxylamine solution (1 g, 0.143 mol  $\text{NH}_2\text{OH}\cdot\text{HCl}$  in 4 ml 10% NaOH) and heated under reflux for 2 h. The solution was concentrated *in vacuo* and extracted with  $\text{CHCl}_3$  (2  $\times$  10 ml). The crude product obtained was recrystallised from  $\text{CHCl}_3$  to give the *oxime* (58b) (0.36 g, 70%) as prisms, m.p. 170—171 °C (Found: C, 65.5; H, 6.7; N, 20.7.  $\text{C}_{15}\text{H}_{18}\text{N}_4\text{O}$  requires C, 65.6; H, 6.7; N, 20.8%);  $\nu_{\text{max}}$  (Nujol) 3380 and 3160  $\text{cm}^{-1}$  (OH stretch);  $\lambda_{\text{max}}$  ( $\text{CHCl}_3$ ) 246 nm (log  $\epsilon$  4.04);  $m/e$  270 (55%).

**7-(5-Nitro-2-pyridyl)-7-azabicyclo[4.3.1]deca-3,8-dien-10-one Phenylhydrazone (59a).**—A solution of the *cycloadduct* (56a) (150 mg, 0.0005 mol) in EtOH (20 ml) was mixed with a clear solution of  $\text{PhNHNH}_2\cdot\text{HCl}$  (0.5 g, 0.003 mol) and NaOAc (1.0 g, 0.01 mol) in  $\text{H}_2\text{O}$  (25 ml). The precipitate formed after stirring the mixture for 1 h at room temperature was collected and washed with  $\text{H}_2\text{O}$  to give the crude *phenylhydrazone* (59a) (174 mg, 96%), as yellow prisms, m.p. 197—198 °C (EtOH) (Found: C, 66.3; H, 5.1.  $\text{C}_{20}\text{H}_{19}\text{N}_5\text{O}_2$  requires C, 66.5; H, 5.3%);  $\nu_{\text{max}}$  (Nujol) 3350 (NH) and 1600  $\text{cm}^{-1}$  (C–N);  $\lambda_{\text{max}}$  ( $\text{CHCl}_3$ ) 345 nm (log  $\epsilon$  3.70);  $m/e$  361 (16%).

**7-(4,6-Dimethylpyrimidin-2-yl)-7-azabicyclo[4.3.1]deca-3,8-dien-10-ol (60b).**—(a) A solution of  $\text{NaBH}_4$  (30 mg) in NaOH solution (20 ml, 0.2N) was added dropwise to a solution of the *cycloadduct* (56b) (0.5 g, 0.0019 mol) in MeOH (25 ml). The reaction mixture was stirred at room temperature for 15 min and then evaporated to dryness. The residue was extracted with  $\text{Et}_2\text{O}$  (3  $\times$  5 ml) to give the *alcohol* (60b) (0.3 g, 61%) as prisms, m.p. 149—150 °C (EtOH) (Found: C, 69.9; H, 7.5; N, 16.2.  $\text{C}_{15}\text{H}_{19}\text{N}_3\text{O}$  requires C, 70.0; H, 7.4; N, 16.4%);  $\nu_{\text{max}}$  (Nujol) 3330  $\text{cm}^{-1}$  (OH);  $\lambda_{\text{max}}$  ( $\text{CHCl}_3$ ) 279 nm (log  $\epsilon$  4.32);  $m/e$  257 (48%).

(b) A solution of the *cycloadduct* (56b) (0.5 g, 0.0019 mol) and NaOH (5 g) in a mixture of EtOH (40 ml) and  $\text{H}_2\text{O}$  (25 ml) was heated under reflux until the *cycloadduct* was completely consumed (t.l.c.,  $\text{CHCl}_3$ ). The solution was then neutralised with glacial AcOH and extracted with  $\text{CHCl}_3$  (2  $\times$  10 ml). The  $\text{CHCl}_3$  extracts were evaporated to dryness *in vacuo* and the crude product purified by preparative t.l.c. ( $\text{CHCl}_3$ ) to give the *alcohol* (60b) (0.3 g, 61%) as prisms, m.p. 149—150 °C (EtOH).

**7-(4,6-Dimethylpyrimidin-2-yl)-10-phenyl-7-azabicyclo[4.3.1]deca-3,8-dien-10-ol (61b).**—A mixture of a few crystals of iodine and magnesium powder (190 mg, 0.0079 mol) in anhydrous  $\text{Et}_2\text{O}$  was heated under reflux. Dry bromo-

benzene (1.23 g, 0.0078 mol) was added dropwise during 1 h. Then a solution of the cycloadduct (56b) (1 g, 0.0039 mol) in Et<sub>2</sub>O (10 ml) was added dropwise. The reaction mixture was heated under reflux for a further 30 min. Ice (10 g) was added to the cooled reaction mixture. The precipitate was removed by filtration and the filtrate extracted with CHCl<sub>3</sub>. The extract was purified by t.l.c. (CHCl<sub>3</sub>) to give the *alcohol* (61b) (1 g, 77%) as needles, m.p. 145–146 °C (EtOH) (Found: C, 75.2; H, 7.1; N, 12.6. C<sub>21</sub>H<sub>23</sub>N<sub>3</sub>O requires C, 75.1; H, 7.0; N, 12.6%);  $\nu_{\text{max}}$  (Nujol) 3330 cm<sup>-1</sup> (OH stretch);  $\lambda_{\text{max}}$  (CHCl<sub>3</sub>) 279 nm (log  $\epsilon$  4.49); *m/e* 333 (94%).

7-(4,6-Dimethylpyrimidin-2-yl)-7-azabicyclo[4.3.1]decan-10-one (62b).—A solution of the cycloadduct (56b) (0.5 g, 0.0019 mol) in EtOH (200 ml) was hydrogenated over palladium-charcoal (100 mg) under 30 lb in<sup>-2</sup> hydrogen pressure at room temperature for 2 days. The course of the reaction was followed by t.l.c. (CHCl<sub>3</sub>). On complete hydrogenation, the catalyst was filtered off and the filtrate evaporated *in vacuo*. The residue was purified by preparative t.l.c. (CHCl<sub>3</sub>) to give the *adduct* (62b) (0.36 g, 73%) as prisms, m.p. 89–90 °C (EtOH) (Found: C, 69.4; H, 8.2; N, 16.1. C<sub>15</sub>H<sub>21</sub>N<sub>3</sub>O requires C, 69.5; H, 8.1; N, 16.2%);  $\nu_{\text{max}}$  (Nujol) 1711 cm<sup>-1</sup> (saturated CO);  $\lambda_{\text{max}}$  (CHCl<sub>3</sub>) 275 nm (log  $\epsilon$  3.63); *m/e* 259 (65%).

7-(4,6-Dimethylpyrimidin-2-yl)-7-azabicyclo[4.3.1]decan-10-one Oxime (63b).—A solution of compound (62b) (0.5 g, 0.0019 mol) in EtOH (50 ml) was mixed with a solution of NH<sub>2</sub>OH·HCl (1 g, 0.0014 mol) and NaOAc (2 g, 0.23 mol) in H<sub>2</sub>O (25 ml). The reaction mixture was heated under reflux for 2 h. Then the solution was concentrated *in vacuo* and extracted with CHCl<sub>3</sub> (2 × 10 ml) to give the *oxime* (63b) (0.35 g, 68%) as prisms, m.p. 170–171 °C (EtOH) (Found: C, 65.4; H, 8.4; N, 20.5. C<sub>15</sub>H<sub>22</sub>N<sub>4</sub>O requires C, 65.7; H, 8.0; N, 20.4%);  $\nu_{\text{max}}$  (Nujol) 3280 and 3360 cm<sup>-1</sup> (OH stretch);  $\lambda_{\text{max}}$  (CHCl<sub>3</sub>) 246 nm (log  $\epsilon$  4.15); *m/e* 274 (50%).

7-(4,6-Dimethylpyrimidin-2-yl)-7-azabicyclo[4.3.1]decan-10-ol (64b).—An ethanolic solution of compound (62b) (0.5 g, 0.0019 mol) in 50 ml EtOH) was mixed with NaOH solution (3 g, in 25 ml H<sub>2</sub>O) and heated under reflux at 100 °C. The course of the reaction was followed by t.l.c. (CHCl<sub>3</sub>) and when all the starting material was consumed the reaction mixture was neutralised (dilute HCl) and extracted with CHCl<sub>3</sub> (2 × 10 ml). The crude product obtained from the CHCl<sub>3</sub> extracts was purified by preparative t.l.c. (CHCl<sub>3</sub>) to give the *alcohol* (64b) (0.3 g, 60%) as prisms, m.p. 139–140 °C [EtOH–Et<sub>2</sub>O (1 : 1)] (Found: C, 66.7; H, 9.0; N, 13.6. C<sub>15</sub>H<sub>23</sub>N<sub>3</sub>O requires C, 66.9; H, 8.9; N, 13.8%);  $\nu_{\text{max}}$  (Nujol) 3280 cm<sup>-1</sup> (OH stretch);  $\lambda_{\text{max}}$  (CHCl<sub>3</sub>) 255 nm (log  $\epsilon$  3.95); *m/e* 261 (98%).

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